BHARATIYA VIDYA BHAVAN'S V.M.PUBLIC SCHOOL, VADODARA
SESSION 2017-18
Sample paper-2

Class: XII Max Marks: 70
Subject: Chemistry Time Allotted: 3 hrs

General Instructions:
1. All questions are compulsory.
2. Marks for each question are indicated against it.
3. Q.No. 1 to 5 are very short questions and carry one mark each.
4. Q.No.6 to 10 are short answer questions of 2 marks each.
5. Q.No.11 to 22 are also short answer questions and carry 3 marks each.
6. Q.No. 23 is value based question and carries 4 marks.
7. Q.No. 24 to 26 are long answer questions and carry 5 marks each.
8. Use log tables if necessary .Use of calculators is not permitted.

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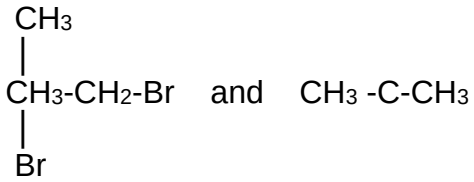
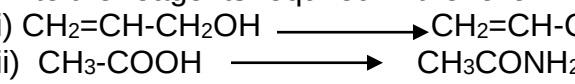
1. What is basiciy of H_3PO_4? 1
2. Which would undergo S_N2 reaction faster in the following pair and why? 1



3. Out of $BaCl_2$ and KCl, which one is more effective in causing coagulation of a negatively charged colloidal sol? Give reason. 1
4. Write the IUPAC name of the given compound. 1

5. A delta is forming at the melting point of sea water .Why? 1
6. Write the reagents required in the following reactions: 2



7. What is the formula of a compound in which the element Y forms ccp lattice and atoms of X occupy 1/3rd of tetrahedral voids? 2
8. What are the transition elements? Write two characteristics of the transition elements. 2
9. What is meant by positive deviation from Raoult's law? Give an example. What is the sign of $\Delta_{mix}H$ for positive deviation? 2

OR

Define azeotropes.What type of azeotrope is formed by positive deviation from Raoult's law? Give an example.

10. a) Following reactions occur at cathode during the electrolysis of aqeous silver chloride solution: 2

$Ag^+(aq) + e^- \rightarrow Ag(s)$ $E^0 = +0.80V$

$H^+(aq) + e^- \rightarrow 1/2H_2(g)$ $E^0 = 0.00V$

On the basis of their standard reduction electrode potential (E^0) values, which reaction is feasible at the cathode and why?

11. 3.9 g of benzoic acid dissolved in 49 g of benzene shows a depression in freezing point

of 1.62 K. Calculate the Van't Hoff factor and predict the nature of solute (associated or dissociated) 3

(Given: Molar Mass of benzoic acid = 122g mol^{-1}, K_f for benzene = 4.9 K kg mol^{-1})

12. (a) Indicate the principle behind the method used for the refining of zinc.
(b) What is the role of silica in the extraction of copper? 3
(c) Which form of the iron is the purest form of commercial iron?

13. An element with molar mass 27 g mol^{-1} forms a cubic unit cell with edge length 3
$4.05X10^{-8}$cm. If its density is 2.7 g cm^{-3}, what is the nature of the cubic unit cell?

14. (a) How would you account for the following: 3
(i) Actinoid contraction is greater than lanthanoid contraction.
(ii) Transition metals form coloured compounds.
(b) Complete the following equation :
$2Mno_4^- + 6H^+ + 5NO_2^- \rightarrow$

15. (a) Draw the geometrical isomers of complex $[Pt(NH_3)_2Cl_2]$ 3
(b) On the basis of crystal field theory, write the electronic configuration for d^4 ion if $\Delta_0 < P$.
(c) Write the hybridization and magnetic behaviour of the complex $[Ni(CO)_4]$.
(At. no. of Ni = 28)

16. Calculate emf of the following cell at 25ºC: 3
Fe | Fe^{2+} (0.001M) || H^+ (0.01M) | H_2(g) (1 bar) | Pt(s)
$E^0(Fe^{2+} | Fe) = -0.44V$ $E^0(H^+ | H_2) = 0.00V$

17. Give reasons for the following: 3
(a) Leather gets hardened after tanning.
(b) Lyophilic sol is more stable than lyophobic sol.
(c) It is necessary to remove CO when ammonia is prepared by Haber's process.

18. Write the names and structures of the monomers of the following polymers: 3
(a) Nylon-6,6 (b) PHBV (c) Neoprene

19. Predict the products of the following reactions: 3
(a) $CH_3—C(=O)—CH_3 \rightarrow ?$
(b) $C_6H_5—CO—CH_3 + NaOH/I_2 \rightarrow ? + ?$
(c) $CH_3COONa + NaoH/CaO \rightarrow ?$

20. How do you convert the following: 3
(a) Phenol to Aniline
(b)Propan-2-ol to 2-methylpropan-2-ol
(c) Aniline to Phenol

OR

(a) Write the mechanism of the following reaction:
$2CH_3CH_2OH \rightarrow CH_3CH_2\text{-}O\text{-}CH_2CH_3$
(b) Write the equation in the acetylation of Salicylic Acid.

21.(a) Which one of the following is a disaccharide : Starch, Maltose, Fructose, Glucose?
(b) What is the difference between fibrous protein and globular protein? 3
(c) Write the name of vitamin whose deficiency causes bones deformities in children.

22. Give reasons: 3
(a) n-Butyl bromide has higher boiling point than t-butyl bromide.
(b) Racemic mixture is optically inactive.
(c) The presence of nitro (NO_2) at o/p positions increases the reactivity of haloarenes towards nucleophilic substitution reaction.

23. Mr. Roy, the principal of one reputed school organized a seminar in which he invited

parents and principals to discuss the serious issue of diabetes and depression in students. They all resolved this issue by strictly banning the junk food in schools and to introduce healthy snacks and drinks like soup, lassi, milk etc. in school canteens. They also decided to make compulsory half and hour physical activities for the students in the morning assembly daily. after six months, Mr. Roy conducted health survey in most of the schools and discovered a tremendous improvement in the health of the students.
After reading the above passage, answer the following: 4

(a) What are the values (at least two) displayed by Mr. Roy?
(b) As a student, how can you spread awareness about this issue?
(c) What are tranquilizers? Give an example.
(d) Why is use of aspartame limited to cold foods and drinks?

24. (a) Account for the following: 5

(i) Acidic character increases from HF to HI.
(ii) There is large difference between the melting and boilinh points of oxygen and sulphur.
(iii) Nitrogen does not form pentahalide.

(b) Draw the structures of the following:

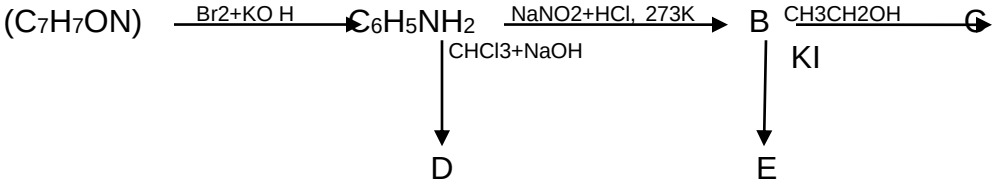
(i) ClF_3 (ii) XeF_4

OR

(a) Which allotrope of phosphorous is more reactive and why?
(b) How the supersonic jet aeroplanes are responsible for the depletion of ozone layer?
(c) F_2 has lower bond dissociation enthalpy than Cl_2. Why?
(d) Which noble gas is used in filling balloons for meteorological observations?
(e) Complete the equations:

$$XeF_2 + PF_5 \rightarrow$$

25. An aromatic compound 'A' of molecular formula C_7H_7ON undergoes a series of reactions as shown below. Write the structures of A, B, C, D and E in the following reactions: 5



OR

a) Write the structures of main products when aniline reacts with the following reagents:
i) Br_2 Water ii) HCl iii) $(CH_3CO)_2O$/pyridine

b) Arrange the following in the increasing order of their boiling point:

$C_2H_5NH_2$, C_2H_5OH, $(CH_3)_3N$

c) Give a simple chemical test to distinguish between the following pair of compounds:

$(CH_3)_2NH$ and $(CH_3)_3N$

26. For the hydrolysis of methyl acetate in aqeous solution, the following results were obtained; 5

t/s	0	30	60
$[CH_3COOCH_3]$/mol L^{-1}	0.60	0.30	0.15

i) Show that it follows pseudo first order reaction, as the concentration of water remains constant.
ii) Calculate the average rate of reaction between the time interval 30 to 60 seconds.

OR

a)For a reaction A+B→P,the rate is given by Rate =k[A] $[B]^2$

i)How is the rate of reaction affected if the concentration of B is doubled?

ii)What is the overall order of reaction if A is present in large excess?

b)A first order reaction takes 30 minutes for 50% completion. Calculate the time required for 90% completion of this reaction.

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